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-part1 **The Tunnel Effect In Chemistry**

The suggestion that quantum-mechanical tunnelling might be a significant factor in some chemical reactions was first made fifty years ago by Hund, very soon after the principles of wave mechanics had been established by de Broglie, Schrodinger and Heisenberg, and similar ideas were put

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In quantum mechanics, these particles can,
with a very small probability, tunnel to the
other side, thus crossing the barrier. Here,
the "ball" could, in a sense, borrow energy
from its surroundings to tunnel through the
wall or "roll over the hill", paying it back
by making the reflected electrons more
energetic than they otherwise would have
been.

Quantum tunnelling - Wikipedia

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