

## Chemistry Review Sheet Unit 7 Answers

Eventually, you will entirely discover a extra experience and endowment by spending more cash. yet when? pull off you resign yourself to that you require to get those all needs with having significantly cash? Why don't you try to get something basic in the beginning? That's something that will lead you to comprehend even more in this area the globe, experience, some places, once history, amusement, and a lot more?

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are representative of each type of problem worked in this unit. For more practice on a given type of problem, refer back to Note Taking Guides, worksheets, and quizzes. 1. Calculate the molar mass for ethane, C<sub>2</sub>H<sub>6</sub>. C<sub>2</sub>H<sub>6</sub> 2(12.0 g) + 6(1.0 g) = 30.0 g 2. Convert 45 g Se to moles of Se. ? mol Se = 45 g Se ! 1 mol Se 79.0 g Se = 0.57 mol Se 3.

~~Review Sheet: Unit 7 Name Vocabulary: Fill in the blanks ...~~

are representative of each type of problem worked in this unit. For more practice on a given type of problem, refer back to Note Taking Guides, worksheets, and quizzes. 1. Calculate the molar mass for ethane, C<sub>2</sub>H<sub>6</sub>. 2. Convert 45 g Se to moles of Se. 3. Convert 4.77 x 10<sup>24</sup> molecules of SO<sub>2</sub> to grams.

~~Review Sheet: Unit 7 Name Vocabulary: Fill in the blanks ...~~

UNIT 7 REVIEW SHEET - EQUILIBRIUM • EQUILIBRIUM is the point at which the rate of the forward reaction is equal to the rate of the reverse reaction in a reversible process. o Reaction is dynamic - still occurring - even though to the eye, it looks like nothing is changing at equilibrium. o Observable/measurable quantities (concentration, pressure, color) are constant at equilibrium. o The amounts of products and reactants are typically not equal at equilibrium!

~~UNIT 7 REVIEW SHEET.pdf UNIT 7 REVIEW SHEET \u2013~~

CHEMISTRY CP Name: \_\_\_\_\_ Per: \_\_\_\_\_ TEST: \_\_\_\_\_ Unit 7 Review Sheet: Thermochemistry KEY Potential Energy Diagrams: Choose the appropriate letter. 1. What is the  $\Delta H$  of the forward reaction? f 2. What is the activation energy of the forward reaction? b 3. What is the  $\Delta H$  of the reverse reaction? f 4.

~~Chemistry Unit 7 Review Sheet KEY OAK PARK USD~~

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Review Sheet: Unit 5 Name \_\_\_\_\_ CHEMISTRY: A Study of Matter © 2004, GPB 5.24a KEY I. For each of the following statements, write "I" for ionic, "C" for covalent

~~CHEMISTRY: A Study of Matter~~

Review Sheet for Chemistry Unit 5 December 11, 2011 31. Balance the following equations by putting the proper coefficients wherever they are needed. a) Sr + O<sub>2</sub> SrO b) Al + H<sub>2</sub>O Al(OH)<sub>3</sub> + H<sub>2</sub> c) Al + O<sub>2</sub> Al<sub>2</sub>O<sub>3</sub> d) C<sub>5</sub>H<sub>12</sub> + O<sub>2</sub> CO<sub>2</sub> + H<sub>2</sub>O e) Na + H<sub>2</sub>O H<sub>2</sub> + NaOH f) NaClO<sub>3</sub> NaCl + O<sub>2</sub> g ...

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