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## 11 Alcohols And Ethers

11.27. Preparation of ethers by acid dehydration of secondary or tertiary alcohols is not a suitable method. Give reason. Ans: Acid catalysed dehydration of primary alcohols to ethers occurs by  $S_N2$  reaction involving nucleophilic attack by the alcohol molecule on the protonated alcohol molecule.

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Chapter 11 - Alcohols, Phenols and Ethers  
Exercise 325 Solution 1 Primary alcohol ' (i), (ii), (iii) Secondary alcohol ' (iv), (v)

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The formation of ethers by dehydration of alcohol is a bimolecular reaction ( $S_N2$ ) involving the attack of an alcohol molecule on a protonated alcohol molecule. In the method, the alkyl group should be unhindered. In case of secondary or tertiary alcohols, the alkyl group is hindered. As a result, elimination dominates substitution.

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Chemistry Chapter-11: Alcohols, Phenols and Ethers. Q 11.1 : Give IUPAC names of the compounds given below : Solution : (i) 2, 2, 4 -Trimethylpentan – 3 – ol (ii) 5 – Ethylheptane – 2, 4 – diol (iii) Butane – 2, 3 – diol (iv) Propane – 1, 2, 3 – triol (v) 2-Methylphenol (vi) 4 – Methyl phenol (vii) 2, 5 – Dimethylphenol

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MP Board Class 12th Chemistry Solutions Chapter 11 Alcohols, Phenols and Ethers Alcohols, Phenols and Ethers NCERT Intext Exercises. Question 1. Classify the following as primary, secondary and tertiary alcohols : Answer: (i) 1 ° , (ii) 1 ° , (iii) 1 ° ,

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(iv) 2°, (v) 2°, (vi) 3°. Question 2.

Identify allylic alcohols in the above examples. Answer:

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In cleavage of ethers, higher alcohol and lower alkyl halide is formed except in tert. butyl methyl ether, in which  $(\text{CH}_3)_3\text{C}^+$  is more stable so  $(\text{CH}_3)_3\text{C-I}$  and  $\text{CH}_3\text{OH}$  is formed. Anisole with HI gives phenol and  $\text{CH}_3\text{I}$  because phenoxide ion is stabilised by resonance.

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Chemistry Notes for class 12 Chapter 11 Alcohols, Phenols and Ethers Alcohols and Phenols Alcohols and phenols are formed when a hydrogen atom in hydrocarbon, aliphatic and aromatic respectively, is replaced by hydroxyl group (-OR group).



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Classification of Alcohols and Phenols In alcohols, -OR group is attached to Sp<sup>3</sup> hybridised carbon.

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Alcohols, Phenols and Ethers Class 12 Notes  
Chemistry Chapter 11. 1. Alcohols and phenols may be classified as monohydric, dihydric, trihydric or polyhydric according to number of hydroxyl groups they contain one, two, three or many respectively in their molecules. 2.

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hydrogen atom in a hydrocarbon, aliphatic

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and aromatic respectively, is replaced by – OH group. The substitution of a hydrogen atom in a hydrocarbon by an alkoxy or aryloxy group (R – O/Ar – O) yields another class of compounds known as ‘ ethers ’ , for example,  $\text{CH}_3\text{OCH}_3$  (dimethyl ether).

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The formation of ethers by dehydration of alcohol is a bimolecular reaction (S N 2)

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involving the attack of an alcohol molecule on a protonated alcohol molecule. In the method, the alkyl group should be unhindered. In case of secondary or tertiary alcohols, the alkyl group is hindered. As a result, elimination dominates over substitution.

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MP Board Class 12th Chemistry Important Questions Chapter 11 Alcohols, Phenols and Ethers Alcohols, Phenols and Ethers Important Questions. Alcohols, Phenols and Ethers Very Short Answer Type Questions. Question 1. What is Ether?  
Answer: Compounds formed by the substitution of hydrogen atom of hydrocarbon by alkoxy group are called Ethers. Question 2.

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